Ionic Compounds

1. Complete the following table.

|  |  |
| --- | --- |
| Name of Compound | Formula |
| (a) beryllium fluoride |  |
| (b) sodium nitride |  |
| (c) calcium sulphide |  |
| (d) aluminum chloride |  |
| (e) lithium oxide |  |
| (f) magnesium nitride |  |
| (g) gallium sulphide |  |
| (h) barium bromide |  |

ANSWER KEY

BLM 2-12, Binary Ionic Compounds

1. (a) BeF2

# (b) Na3N

# (c) CaS

# (d) AlCl3

# (e) Li2O

# (f) Mg3N2

# (g) GaS

# (h) BaBr2