Souths are a class of ionic compounds that can be formed during a reaction of an acid and a base

AUD- BASE NEUTRALIZATION

-occurs when an acid and a base react to form a sout and water

ex. HCl + NaOH -> NaCl + H2O ac'd base salt water

METAL OXIDES and NON-METAL OXIDES

oxides include one oxygen atom along w/

metal oxides compounds that contain a metal combined w/ Oxygen

ex. $Na_2O + H_2O \rightarrow NaOH$

n<u>on-metal oxides</u> compounds that contain a non-metal combined with oxygen.

 $(x. 50_2 + H_2O \rightarrow H_2SO_3)$

ACIDS and METALS

* most reactive metals on LEFT side of P.T. # Lower is more reactive than higher up on P.T.

- when metals react w/ acid, release hydrogen gas (H2)

ex. HCl + Mg -> Mg Cl, + Hz (9)

acid metal salt hydrogen

gas

magnesium

ACIDS and CARBONATES

- carbonate vocks react with acids.

ex. H2SOy + (a(O3 → (aSOy + H2O + CO2

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